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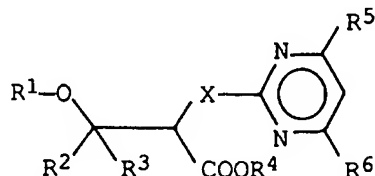
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DE FR GB IT(71) Applicant: **UBE INDUSTRIES, LTD.**
12-32, Nishihonmachi 1-chome
Ube-shi, Yamaguchi-ken 755(JP)(72) Inventor: **Harada, Katsumasa, c/o Ube**
Research Lab.
UBE Industries, Ltd., 1978-5, Oaza Kogushi
Ube-Shi, Yamaguchi-ken(JP)
Inventor: **Abe, Takaaki, c/o Ube Research Lab.**
UBE Industries, Ltd., 1978-5, Oaza Kogushi
Ube-Shi, Yamaguchi-ken(JP)Inventor: **Akiyoshi, Yuji, c/o Ube Research**
Lab.**UBE Industries, Ltd., 1978-5, Oaza Kogushi**
Ube-Shi, Yamaguchi-ken(JP)Inventor: **Shiraishi, Hiroshi, c/o Ube Research**
Lab.**UBE Industries, Ltd., 1978-5, Oaza Kogushi**
Ube-Shi, Yamaguchi-ken(JP)Inventor: **Yamamoto, Kaoru, c/o Ube Research**
Lab.**UBE Industries, Ltd., 1978-5, Oaza Kogushi**
Ube-Shi, Yamaguchi-ken(JP)(74) Representative: **Hansen, Bernd, Dr.**
Dipl.-Chem. et al
Hoffmann, Eitle & Partner Patent- und
Rechtsanwälte Arabellastrasse 4 Postfach
81 04 20
W-8000 München 81(DE)(54) **3-Alkoxyalkanoic acid derivative, process for preparing the same and herbicide using the same.**

(57) Disclosed is a 3-alkoxyalkanoic acid compound represented by the following formula (I):



(I)

wherein R¹ represents a lower alkyl group, a lower alkenyl group, a lower alkynyl group, a cycloalkyl group, a halo-lower alkyl group or a cyano-lower alkyl group; R² represents hydrogen atom or a lower alkyl group; R³ represents a lower alkyl group or a cycloalkyl group; or R² and R³ are mutually bonded to represent a cycloalkyl group; R⁴ represents hydrogen atom, a lower alkyl group or a lower alkynyl group; R⁵ represents a lower alkoxy group, a lower alkyl group, a halogen atom or a halo-lower alkyl group; R⁶ represents a lower alkoxy group or a lower alkyl group; and X represents oxygen atom or sulfur atom, or an alkali addition salt thereof,

processes for preparing the same and herbicides containing the same as an active ingredient(s).

BACKGROUND OF THE INVENTION

This invention relates to a herbicide containing a novel 3-alkoxyalkanoic acid derivative as an active ingredient.

Many herbicides have heretofore been developed for promoting labor-saving of farm practices and increase in productivity of crops. Conventional herbicides are, however, not sufficient in herbicidal effect, and also not sufficiently satisfactory in the point of safety to creatures. Thus, in order to solve these problems, development of a novel herbicide has been demanded.

As a compound similar to the novel 3-alkoxyalkanoic acid derivative of the present invention, there has been known, for example, compounds disclosed in Japanese Provisional Patent Publication No. 85262/1990, and it has been also known that these compounds have herbicidal activities.

However, in the above patent publication, there are descriptions about 3-hydroxyalkanoic acid derivatives and 4-hydroxyalkanoic acid derivatives, but there is no description about a 3-alkoxyalkanoic acid derivative.

Thus, it has been not known that a 3-alkoxyalkanoic acid derivative has herbicidal effects, and further has more excellent herbicidal effect than the compounds disclosed in the above patent publication.

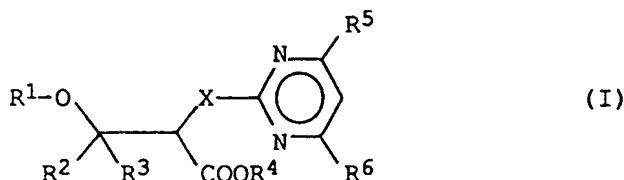
SUMMARY OF THE INVENTION

An object of the present invention is to provide a novel 3-alkoxyalkanoic acid derivative or an alkali addition salt thereof, a process for preparing the same and a herbicide which contains said compound as an active ingredient.

The present inventors have studied intensively in order to solve the above problems, and consequently found that a novel 3-alkoxyalkanoic acid derivative shows more excellent herbicidal effect than the compounds disclosed in Japanese Provisional Patent Publication No. 85262/1990 (a 3-hydroxyalkanoic acid derivative and a 4-hydroxyalkanoic acid derivative), to accomplish the present invention.

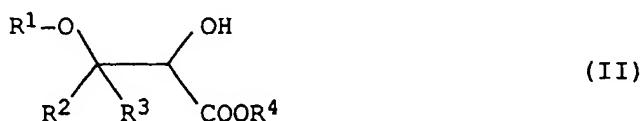
The present invention is described below.

That is, the first invention is concerned to a 3-alkoxyalkanoic acid derivative represented by the following formula (I):



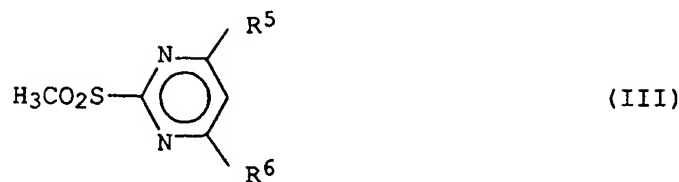
wherein R¹ represents a lower alkyl group, a lower alkenyl group, a lower alkynyl group, a cycloalkyl group, a halo-lower alkyl group or a cyano-lower alkyl group; R² represents hydrogen atom or a lower alkyl group; R³ represents a lower alkyl group or a cycloalkyl group; or R² and R³ are mutually bonded to represent a cycloalkyl group; R⁴ represents hydrogen atom, a lower alkyl group or a lower alkynyl group; R⁵ represents a lower alkoxy group, a lower alkyl group, a halogen atom or a halo-lower alkyl group; R⁶ represents a lower alkoxy group or a lower alkyl group; and X represents oxygen atom or sulfur atom, or an alkali addition salt thereof.

The second invention is concerned to a process for preparing the 3-alkoxyalkanoic acid derivative represented by the above formula (I), which comprises reacting a compound represented by the following formula (II):



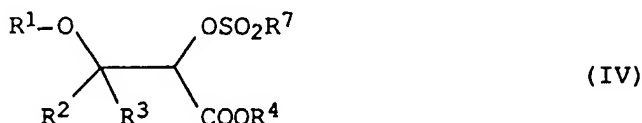
wherein R¹, R², R³ and R⁴ each have the same meanings as defined above,

with a compound represented by the following formula (III):

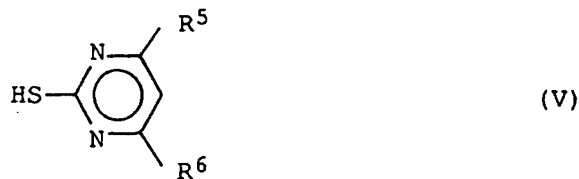


wherein R⁵ and R⁶ each have the same meanings as defined above.

The third invention is concerned to a process for preparing the 3-alkoxyalkanoic acid derivative represented by the above formula (I), which comprises reacting a compound represented by the following formula (IV):

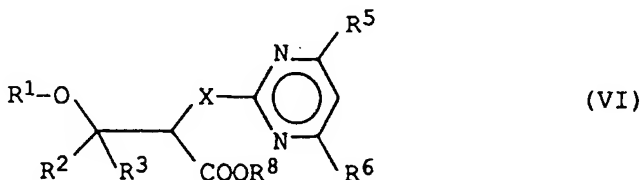


wherein R¹, R², R³ and R⁴ each have the same meanings as defined above; and R⁷ represents a lower alkyl group, a substituted phenyl group or a halo-lower alkyl group, with a compound represented by the following formula (V):



wherein R⁵ and R⁶ each have the same meanings as defined above.

The fourth invention is concerned to a process for preparing the 3-alkoxyalkanoic acid derivative represented by the above formula (I), which comprises hydrolyzing a compound represented by the following formula (VI):



wherein R¹, R², R³, R⁵, R⁶ and X each have the same meanings as defined above; and R⁸ represents a lower alkyl group or a lower alkynyl group,

The fifth invention is concerned to a herbicide comprising the 3-alkoxyalkanoic acid derivative represented by the above formula (I) or an alkali addition salt thereof as an active ingredient.

DESCRIPTION OF THE PREFERRED EMBODIMENTS

In the following, the present invention is explained in detail.

In the novel 3-alkoxyalkanoic acid derivative (I) which is the desired compound of the present invention and the compounds (II) to (VI) which are starting materials thereof, R¹, R², R³, R⁴, R⁵, R⁶, R⁷, R⁸ and X are as described below.

5 As R¹, there may be mentioned a lower alkyl group, a lower alkenyl group, a lower alkynyl group, a cycloalkyl group, a halo-lower alkyl group and a cyano-lower alkyl group, preferably a lower alkyl group (e.g. a straight or branched alkyl group having 1 to 6 carbon atoms which may be substituted by a cycloalkyl group having 3 to 8 carbon atoms), a lower alkenyl group (e.g. a straight or branched alkenyl group having 2 to 6 carbon atoms), a lower alkynyl group (e.g. a straight or branched alkynyl group having
10 2 to 6 carbon atoms), a cycloalkyl group (e.g. a cycloalkyl group having 3 to 8 carbon atoms), a halo-lower alkyl group (e.g. a straight or branched haloalkyl group having 2 to 6 carbon atoms) and a cyano-lower alkyl group (e.g. a straight or branched cyanoalkyl group having 2 to 6 carbon atoms). The lower alkyl group is more preferably an alkyl group having 1 to 4 carbon atoms (e.g. methyl group, ethyl group, n-propyl group, i-propyl group, n-butyl group, i-butyl group, sec-butyl group and t-butyl group) or a straight or branched
15 alkyl group having 1 to 3 carbon atoms which may be substituted by a cycloalkyl group having 3 to 5 carbon atoms (e.g. cyclopropylmethyl group and cyclobutylmethyl group), the lower alkenyl group is more preferably a straight or branched alkenyl group having 2 to 5 carbon atoms (e.g. propenyl group), the lower alkynyl group is more preferably a straight or branched alkynyl group having 2 to 5 carbon atoms (e.g. propynyl group), the cycloalkyl group is more preferably a cycloalkyl group having 3 to 6 carbon atoms
20 (e.g. cyclopentyl group), the halo-lower alkyl group is more preferably a straight or branched halo-lower alkyl group having 2 to 4 carbon atoms (e.g. 2-chloroethyl group and 2-trifluoroethyl group) and the cyano-lower alkyl group is more preferably a straight or branched cyanoalkyl group having 2 to 4 carbon atoms (e.g. 2-cyanoethyl group).

As R², there may be mentioned hydrogen atom and a lower alkyl group, and the lower alkyl group is
25 preferably a straight or branched alkyl group having 1 to 5 carbon atoms (e.g. those having 1 to 5 carbon atoms as described above).

As R³, there may be mentioned a lower alkyl group and a cycloalkyl group, preferably a lower alkyl group (e.g. a straight or branched alkyl group having 1 to 6 carbon atoms), and the lower alkyl group is preferably a straight or branched alkyl group having 1 to 5 carbon atoms (e.g. those having 1 to 5 carbon
30 atoms as described above).

Or else, R² and R³ may be mutually bonded to form a cycloalkyl group (e.g. a cycloalkyl group having 3 to 8 carbon atoms), and the cycloalkyl group is more preferably a cycloalkyl group having 3 to 6 carbon atoms (e.g. cyclopentyl group and cyclohexyl group).

As R⁴, there may be mentioned hydrogen atom, a lower alkyl group and a lower alkynyl group. The
35 lower alkyl group is preferably a straight or branched alkyl group having 1 to 5 carbon atoms (e.g. those having 1 to 5 carbon atoms as described above), and the lower alkynyl group is more preferably a straight or branched alkynyl group having 2 to 5 carbon atoms (e.g. propynyl group).

As R⁵, there may be mentioned a lower alkoxy group, a lower alkyl group, a halogen atom and a halo-lower alkyl group.

40 The lower alkoxy group is preferably a straight or branched alkoxy group having 1 to 5 carbon atoms (e.g. methoxy group, ethoxy group, isopropoxy group and propoxy group), more preferably methoxy group. The lower alkyl group is preferably a straight or branched alkyl group having 1 to 5 carbon atoms (e.g. those having 1 to 5 carbon atoms as described above), more preferably methyl group. The halogen atom (e.g. fluorine atom, chlorine atom, bromine atom and iodine atom) is preferably chlorine atom. The
45 halo-lower alkyl group is preferably a straight or branched alkyl group having 1 to 5 carbon atoms (e.g. trifluoromethyl group), more preferably trifluoromethyl group.

As R⁶, there may be mentioned a lower alkoxy group and a lower alkyl group. The lower alkoxy group is preferably a straight or branched alkoxy group having 1 to 5 carbon atoms (e.g. those having 1 to 5 carbon atoms as described above), more preferably methoxy group. The lower alkyl group is preferably a
50 straight or branched alkyl group having 1 to 5 carbon atoms (e.g. those having 1 to 5 carbon atoms as described above), more preferably methyl group.

As R⁷, there may be mentioned a lower alkyl group, a substituted phenyl group and a halo-lower alkyl group (e.g. those as described above).

As R⁸, there may be mentioned a lower alkyl group (e.g. those having 1 to 5 carbon atoms as
55 described above) and a lower alkynyl group (e.g. those having 2 to 5 carbon atoms as described above).

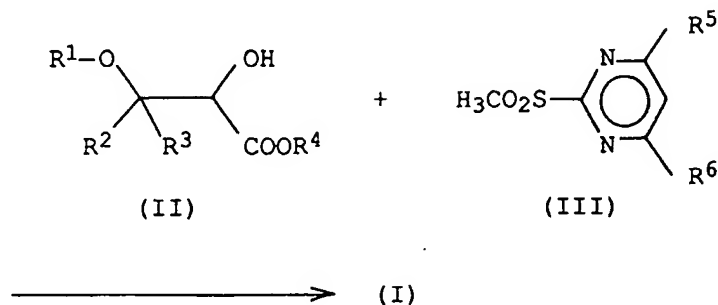
As X, there may be mentioned oxygen atom and sulfur atom.

The desired compound (I) can form an alkali addition salt easily with -COO⁻ by using an alkali metal (e.g. sodium or potassium) or an organic amine (e.g. monosubstituted or disubstituted alkylamine).

The novel 3-alkoxyalkanoic acid derivative (I) which is a desired compound may include an optical isomer based on an asymmetric carbon atom.

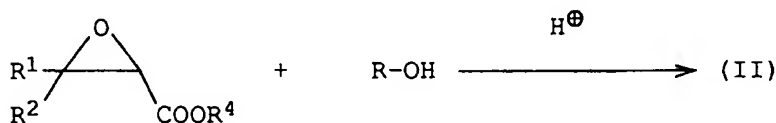
The compound (I) can be prepared by, for example, Preparation method 1 to Preparation method 3 shown below.

(Preparation method 1)



In general, the compound (I) is preferably prepared by reacting the starting compound (II) with the starting compound (III) in a solvent in the presence of a base, but it can be also obtained by reacting the starting compounds (II) and (III) by melting under heat in the absence of a solvent.

The compound (II) can be prepared easily by, for example, reacting epoxyalkanoates prepared according to the method described in "Tetrahedron Letter", No. 36, p. 3761 (1972) or "Org. Syn.", IV, p. 459 in an alcohol in the presence of an acid catalyst (sulfuric acid or paratoluenesulfonic acid) as shown below.



wherein R¹, R² and R⁴ each have the same meanings as defined above; and R represents an alkyl group.

As the compound (II), there may be mentioned, for example, the respective compounds (referred to as Compounds (II)₁ to (II)₁₇₆) comprising the respective kinds of substituted groups corresponding to Compounds No. 1 to No. 172 shown in Table 1, preferably Compounds Nos. 1, 2, 6, 8, 10, 12, 14, 16, 18, 20, 22, 23, 25, 27, 29, 31, 32, 36, 38, 40, 42, 44, 46, 48, 50, 51, 55, 57, 59, 61, 63, 65, 67, 69, 71, 73, 74, 76, 78, 79, 80, 81, 84, 85, 86, 89, 91, 93, 95, 97, 98, 100, 101, 103, 105, 106, 110, 114, 116, 118, 119, 123, 124, 125, 127, 128, 129, 131, 133, 135, 137, 139, 141, 143, 144, 146, 152, 154, 159, 162, 164, 166, 168, 170 and 174 (for example, the compound (II) corresponding to Compound No. 1 is referred to as Compound (II)₁, and this Compound (II)₁ means a compound wherein R¹ is CH₃, R² is H, R³ is CH₃ and R⁴ is CH₃ in the compound (II)).

The compound (III) can be prepared easily according to, for example, the method disclosed in Japanese Provisional Patent Publication No. 23870/1988.

As the compound (III), there may be mentioned, for example, the respective compounds (referred to as Compounds (III)₁ to (III)₁₇₆) comprising the respective kinds of substituted groups corresponding to Compounds No. 1 to No. 176 shown in Table 1, preferably Compounds Nos. 1, 2, 6, 8, 10, 12, 14, 16, 18, 20, 22, 23, 25, 27, 29, 31, 32, 36, 38, 40, 42, 44, 46, 48, 50, 51, 55, 57, 59, 61, 63, 65, 67, 69, 71, 73, 74, 76, 78, 79, 80, 81, 84, 85, 86, 89, 91, 93, 95, 97, 98, 100, 101, 103, 105, 106, 110, 114, 116, 118, 119, 123, 124, 125, 127, 128, 129, 131, 133, 135, 137, 139, 141, 143, 144, 146, 152, 154, 159, 162, 164, 166, 168, 170 and 174 (for example, the compound (III) corresponding to Compound No. 1 is referred to as Compound (III)₁, and this Compound (III)₁ means a compound wherein R⁵ and R⁶ are each OCH₃ in the compound (III)).

The solvent is not particularly limited so long as it does not participate in the present reaction directly, and may include, for example, chlorinated or unchlorinated aromatic or aliphatic hydrocarbons such as benzene, toluene, xylene, methylnaphthalene, chlorobenzene, dichlorobenzene, methylene chloride, chloroform and dichloroethane; ethers such as diethyl ether, tetrahydrofuran and dioxane; ketones such as acetone and methyl ethyl ketone or hydrates thereof; amides such as N,N-dimethylformamide and N,N-dimethylacetamide; 1,3-dimethyl-2-imidazolidinone; dimethyl sulfoxide; and a mixture of the above solvents.

As the base, there may be mentioned, for example, organic bases such as triethylamine, pyridine and N,N-diethylaniline; and inorganic bases such as sodium amide, sodium hydroxide, potassium hydroxide, potassium carbonate, sodium carbonate and sodium hydride.

The reaction for preparing the compound (I) can be carried out at a reaction concentration of 5 to 100 %.

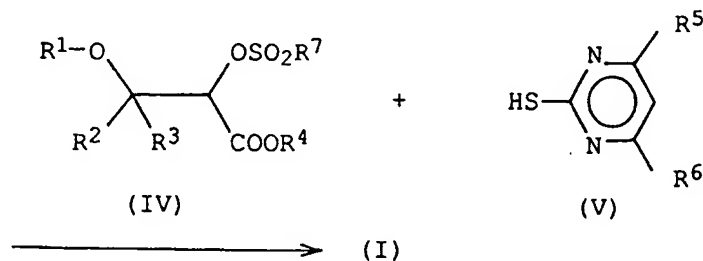
In the preparation method, the ratio of the starting compound (II) to the starting compound (III) to be used is 0.5 to 2 mole, preferably 1 to 1.5 mole of the starting compound (III) per mole of the starting compound (II).

The reaction temperature is not particularly limited so long as it is the boiling point of a solvent used or lower. However, the reaction can be carried out generally at room temperature or higher, and it is preferred to carry out the reaction by heating to shorten the reaction time.

The reaction time varies depending on the above concentration and temperature, but may be generally 1 to 12 hours.

As the compound (I), there may be mentioned, for example, the respective compounds (referred to as Compounds 1 to 176) comprising the respective kinds of substituted groups corresponding to Compounds No. 1 to No. 176 shown in Table 1, preferably Compounds Nos. 1, 2, 6, 8, 10, 12, 14, 16, 18, 20, 22, 23, 25, 27, 29, 31, 32, 36, 38, 40, 42, 44, 46, 48, 50, 51, 55, 57, 59, 61, 63, 65, 67, 69, 71, 73, 74, 76, 78, 79, 80, 81, 84, 85, 86, 89, 91, 93, 95, 97, 98, 100, 101, 103, 105, 106, 110, 114, 116, 118, 119, 123, 124, 125, 127, 128, 129, 131, 133, 135, 137, 139, 141, 143, 144, 146, 152, 154, 159, 162, 164, 166, 168, 170 and 174 (for example, the compound (I) corresponding to Compound No. 1 is referred to as Compound 1, and this Compound 1 means a compound wherein R¹ is CH₃, R² is H, R³ is CH₃, R⁴ is CH₃, R⁵ and R⁶ are each OCH₃ and X is O in the compound (I)).

(Preparation method 2)



wherein R¹, R², R³, R⁴, R⁵, R⁶ and R⁷ each have the same meanings as defined above.

The compound (I) can be obtained by carrying out the reaction in the same manner as in Preparation method 1 except for using the starting compound (IV) in place of the starting compound (II) and using the starting compound (V) in place of the starting compound (III).

As the compound (IV), there may be mentioned, for example, the respective compounds (referred to as Compounds (IV)₁ to (IV)₁₇₆) comprising the respective kinds of substituted groups corresponding to Compounds No. 1 to No. 176 shown in Table 1, preferably Compounds Nos. 4, 34, 35, 53, 54, 88, 108, 109, 113, 121, 122, 156, 157, 158, 161, 169, 172, 173 and 176 (for example, the compound (IV) corresponding to Compound No. 4 is referred to as Compound (IV)₄, and this Compound (IV)₄ means a compound wherein R¹ is CH₃, R² is H, R³ is CH₃, R⁴ is C₂H₅ and R⁷ is a substituted phenyl group in the compound (IV)).

As the compound (V), there may be mentioned, for example, the respective compounds (referred to as Compounds (V)₁ to (V)₁₇₆) comprising the respective kinds of substituted groups corresponding to Compounds No. 1 to No. 176 shown in Table 1, preferably Compounds Nos. 4, 34, 35, 53, 54, 88, 108, 109, 113, 121, 122, 156, 157, 158, 161, 169, 172, 173 and 176 (for example, the compound (V) corresponding to

Compound No. 4 is referred to as Compound (V)₄, and this Compound (V)₄ means a compound wherein R⁵ and R⁶ are each OCH₃ in the compound (V).

The starting compound (V) to be used in the present invention can be prepared easily by, for example, adding sodium hydrosulfide to a corresponding 2-methylsulfonylpyrimidine derivative and stirring the mixture in water or an alcohol under heating.

The reaction for preparing the compound (I) can be carried out at a reaction concentration of 5 to 80 %.

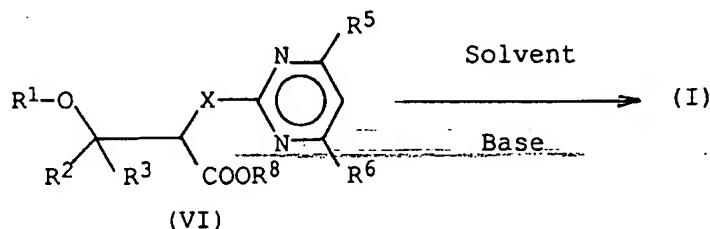
In the preparation method, the ratio of the starting compound (IV) to the starting compound (V) to be used is 0.5 to 2 mole, preferably 1 to 1.5 mole of the starting compound (V) per mole of the starting compound (IV).

As the solvent and base, those described in Preparation method 1 can be used.

The reaction temperature is not particularly limited so long as it is the boiling point of a solvent used or lower. However, the reaction can be carried out generally at room temperature or higher, and the temperature range is preferably 10 to 80 °C, more preferably 20 to 50 °C.

The reaction time varies depending on the above concentration and temperature, but may be generally 1 to 12 hours.

(Preparation method 3)



wherein R¹, R², R³, R⁵, R⁶, R⁸ and X each have the same meanings as defined above.

The compound (I) can be generally prepared by reacting the starting compound (VI) in a solvent in the presence of a base.

As the compound (VI), there may be mentioned, for example, the respective compounds (referred to as Compounds 1 to 176) comprising the respective kinds of substituted groups corresponding to Compounds No. 1 to No. 176 shown in Table 1, preferably Compounds Nos. 3, 7, 9, 11, 13, 15, 17, 19, 21, 24, 26, 28, 30, 33, 37, 39, 41, 43, 45, 47, 49, 52, 56, 58, 60, 62, 64, 66, 68, 70, 72, 75, 77, 82, 87, 90, 92, 94, 96, 99, 102, 104, 107, 111, 115, 117, 120, 126, 130, 132, 134, 136, 138, 140, 142, 145, 147, 153, 155, 160, 163, 165, 167, 171 and 175 (for example, the compound (VI) corresponding to Compound No. 3 is referred to as Compound (VI)₃, and this Compound (VI)₃ means a compound wherein R¹ is CH₃, R² is H, R³ is CH₃, R⁴ is C₂H₅, R⁵ and R⁶ are each OCH₃ and X is O in the compound (VI)).

As the solvent, there may be mentioned alcohols such as methanol and ethanol; ethers such as 1,4-dioxane and tetrahydrofuran; ketones such as acetone and methyl ethyl ketone; water; and a mixture of the above solvents.

As the base, there may be mentioned sodium hydroxide and potassium hydroxide.

The reaction temperature is not particularly limited so long as it is the boiling point of a solvent used or lower, but the reaction can be carried out generally at room temperature or higher and it is preferred to carry out the reaction in the range of 10 to 80 °C, preferably 20 to 50 °C.

The reaction time varies depending on the following concentration and temperature, but may be generally 1 to 12 hours.

The reaction is generally carried out under the concentration of 5 to 60 %.

An alkali addition salt of the compound (I) can be obtained by reacting the compound (I) with a monoequivalent or diequivalent alkali metal or an organic amine in an organic solvent (e.g. an ether or an alcohol) or water at room temperature for 1 to 24 hours.

The herbicide containing the compound (I) or an alkali addition salt thereof as an active ingredient has high selectivity and also shows excellent herbicidal effect.

That is, the herbicide of the present invention shows excellent herbicidal effect on annual weeds and perennial weeds grown in paddy fields and upland fields, and its herbicidal effect is particularly remarkable in annual grass weeds (e.g. crabgrass (manna-grass), barnyardgrass and foxtail (green panicum)), annual

broad-leaved weeds (e.g. morning glory, common lambsquarter (white goosefoot), livid amaranthus and velvetleaf) and perennial weeds (e.g. Johnson grass).

The herbicide of the present invention shows excellent herbicidal effect on the weeds described above, but does not give chemical damage on field crops (e.g. cotton) at a concentration for such a treatment.

5 The herbicide of the present invention contains the compound (I) or at least one alkali addition salt thereof as an active ingredient(s).

The compound (I) can be used singly, but may be preferably used by mixing with a carrier, a surfactant, a dispersant and an auxiliary (for example, prepared as a composition such as a dust, an emulsion, a fine granule, a granule, a wettable powder, an oily suspension and an aerosol) according to a
10 conventional method.

As the carrier, there may be mentioned, for example, a solid carrier such as talc, mica, bentonite, clay, kaolin, diatomaceous earth, white carbon, vermiculite, dolomite, zeolite, slaked lime, siliceous sand, silicic anhydride, ammonium sulfate, urea, wood powder, starch and cellulose; a liquid carrier such as hydrocarbons (kerosine and mineral oil), aromatic hydrocarbons (benzene, toluene and xylene), chlorinated hydrocarbons (chloroform and carbon tetrachloride), ethers (dioxane and tetrahydrofuran), ketones (acetone, cyclohexanone and isophorone), esters (ethyl acetate, ethylene glycol acetate and dibutyl maleate), alcohols (methanol, n-hexanol and ethylene glycol), polar solvents (dimethylformamide and dimethylsulfoxide) and water; and a gas carrier such as air, nitrogen, carbonic acid gas and freon (in the case of a gas carrier, mixed spray can be carried out).

20 As the surfactant which can be used for improving attachment of the present herbicide to and absorption thereof in plants, and improving characteristics such as dispersion, emulsification and spreading of the herbicide, there may be mentioned nonionic, anionic, cationic or amphoteric surfactants (e.g. alcohol sulfates, alkylsulfonates, lignin sulfonates and polyoxyethylene glycol ethers) Further, for improving properties of preparation, carboxymethyl cellulose, polyethylene glycol or gum arabic can be used as an
25 auxiliary.

In preparation of the present herbicide, in addition to the above carrier, surfactant, dispersant and auxiliary, other agricultural chemicals (a fungicide and an insecticide), a fertilizer and a soil conditioner can be used singly or in a suitable combination, respectively, depending on the respective purposes.

When the compound (I) of the present invention is made into preparations, the concentration of the
30 active ingredient is generally 1 to 50 % by weight in an emulsion, generally 0.3 to 25 % by weight in a dust, generally 1 to 90 % by weight in a wettable powder, generally 0.5 to 5 % by weight in a granule, generally 0.5 to 5 % by weight in an oily dispersion, and generally 0.1 to 5 % by weight in an aerosol.

These preparations can be provided for various uses by diluting them to have a suitable concentration and spraying them to stems and/or leaves of plants, soil and paddy field surface, or by applying them
35 directly thereto, depending on the respective purposes.

EXAMPLES

The present invention is described in detail by referring to Examples, but the scope of the present
40 invention is not limited by these Examples.

Example 1

(1) Synthesis of ethyl 2-(4,6-dimethoxypyrimidin-2-yl)oxy-3-ethoxy-3-methylbutanoate (Compound 50)

45 The compound (I) of the present invention was synthesized according to the method described in (Preparation method 1).

That is, to ethanol (40 ml) was added ethyl 2,3-epoxy-3-methylbutanoate (14.4 g, 0.1 mol) and then sulfuric acid (0.5 ml) was added as a catalyst, and the mixture was stirred at 30 °C for 2 hours.

50 Subsequently, excessive ethanol was removed under reduced pressure to obtain ethyl 3-ethoxy-2-hydroxy-3-methylbutanoate. After the product obtained and 4,6-dimethoxy-2-methylsulfonylpyrimidine (21.8 g, 0.1 mol) were added to N,N-dimethylformamide (150 ml) and further anhydrous potassium carbonate (16.6 g, 0.12 mol) was added, the mixture was stirred at 60 °C for 3 hours. The resulting reaction mixture was added to water and extracted with ethyl acetate. The ethyl acetate layer was washed with a saturated
55 saline solution and dried over sodium sulfate, and ethyl acetate was removed under reduced pressure. The residue obtained was isolated by column chromatography (Wako gel C-200 (trade name, produced by Wako Junyaku K.K.), eluted by toluene:ethyl acetate = 4:1) to obtain 26.5 g (yield: 81 %) of the title compound as a white powder.

(2) Synthesis of 2-(4,6-dimethoxypyrimidin-2-yl)oxy-3-ethoxy-3-methylbutanoic acid (Compound 52)

The title compound (I) was synthesized according to the method described in (Preparation method 1).

That is, to Compound 50 (32.8 g, 0.1 mol) prepared in the above (1) dissolved in ethanol (100 ml) was added 5N sodium hydroxide (40 ml), and the mixture was stirred for one hour. Subsequently, ethanol was removed under reduced pressure, 5N hydrochloric acid (50 ml) was added to the residue obtained, and the residue was extracted with chloroform. The chloroform layer was washed with water and dried, and then chloroform was removed under reduced pressure to obtain 25.5 g (yield: 85.0 %) of the title compound as white crystals.

(3) Synthesis of sodium 2-(4,6-dimethoxypyrimidin-2-yl)oxy-3-ethoxy-3-methylbutanoate (Compound 148)

To Compound 52 (30.0 g, 0.1 mol) prepared in the above (2) dissolved in methanol (100 ml) was added sodium methoxide (19.3 g, 28 % methanol solution), and the mixture was stirred for 10 minutes. Subsequently, methanol was removed under reduced pressure to obtain 32 g of the title compound as white crystals.

(4) Synthesis of ethyl 2-(4,6-dimethoxypyrimidin-2-yl)thio-3-(isopropoxy)butanoate (Compound 88)

The title compound (I) was synthesized according to the method described in (Preparation method 2).

That is, to ethyl 2,3-epoxybutanoate (13.0 g, 0.1 mol) dissolved in isopropanol (30 ml) was added sulfuric acid (0.5 ml), and the mixture was stirred at 60 °C for 5 hours. Subsequently, excessive isopropanol was removed under reduced pressure to obtain ethyl 3-(isopropoxy)-2-hydroxybutanoate. The product obtained and p-toluenesulfonyl-chloride (20.9 g, 0.11 mol) were added to pyridine (50 ml), and the mixture was stirred at room temperature for 5 hours. Subsequently, water was added to the reaction mixture, and the oily product liberated was extracted with toluene. After the toluene layer was washed with water and dried over sodium sulfate, toluene was removed under reduced pressure. The oily product obtained was isolated by silica gel column chromatography (Wako gel C-200 (trade name, produced by Wako Junyaku K.K.), eluted by hexane: ethyl acetate = 9:1) to obtain ethyl 3-(isopropoxy)-2-p-toluenesulfonyloxybutanoate (27.5 g).

After this compound, ethyl 3-(isopropoxy)-2-p-toluenesulfonyloxybutanoate (27.5 g, 0.08 mol) and 4,6-dimethoxy-2-mercaptopyrimidine (13.7 g, 0.08 mol) were added to N,N-dimethylformamide (80 ml) and further anhydrous potassium carbonate (11.1 g) was added thereto, the mixture was stirred at room temperature for 5 hours. Subsequently, water was added to the reaction mixture, the mixture was extracted with ethyl acetate, washed with water and dried, and ethyl acetate was removed under reduced pressure. The residue obtained was isolated by column chromatography (Wako gel C-200 (trade name, produced by Wako Junyaku K.K.), eluted by hexane:ethyl acetate = 8:1) to obtain 15.0 g (yield: 54.5 %) of the title compound as a colorless oily product.

(5) Synthesis of 2-(4,6-dimethoxypyrimidin-2-yl)thio-3-methoxy-3-methylbutanoic acid (Compound 21)

The title compound (I) was synthesized according to the method described in (Preparation method 2).

That is, after 3-methoxy-3-methyl-2-methylsulfonyloxybutanoic acid (22.6 g, 0.1 mol) and 4,6-dimethoxy-2-mercaptopyrimidine (17.3 g, 0.1 mol) were added to N,N-dimethylformamide (200 ml) and further anhydrous potassium carbonate (13.8 g) was added, the mixture was stirred at 50 °C for 2 hours. The resulting reaction mixture was added to a 1N hydrochloric acid aqueous solution (300 ml), and extracted with ethyl acetate. The ethyl acetate layer was washed with a saturated saline solution and dried over magnesium sulfate, and ethyl acetate was removed under reduced pressure. The crystals obtained were washed with n-hexane to obtain 24.1 g (yield: 80 %) of the title compound.

(6) Synthesis of ethyl 2-(4,6-dimethoxypyrimidin-2-yl)oxy-3-(n-propoxy)butanoate (Compound 79)

The title compound (I) was synthesized according to the method described in (Preparation method 1).

That is, after ethyl 2,3-epoxy-3-methylbutanoate (14.4 g, 0.1 mol) was added to n-propanol (40 ml) and sulfuric acid (0.5 ml) was added as a catalyst, the mixture was stirred at 30 °C for 4 hours.

Subsequently, excessive n-propanol was removed under reduced pressure to obtain ethyl 2-hydroxy-3-methyl(n-propoxy)-butanoate. After the product obtained and 4,6-dimethoxy-2-methylsulfonylpyrimidine (21.8 g, 0.01 mol) were added to acetone (150 ml) and further anhydrous potassium carbonate (16.6 g, 0.12

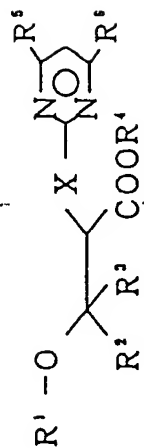
mol) was added, the mixture was refluxed under heating. The resulting reaction mixture was added to water and extracted with ethyl acetate. The ethyl acetate layer was washed with a saturated saline solution and dried over sodium sulfate, and ethyl acetate was removed under reduced pressure. The residue obtained was isolated by column chromatography (Wako gel C-200 (trade name, produced by Wako Junyaku K.K.), eluted by toluene:ethyl acetate = 4:1) to obtain 27.0 g (yield: 80 %) of the title compound as a white powder.

(7) Syntheses of other compounds (I) in Table 1

In the same manner as in either of the synthetic methods (1) to (6), the title compounds (I) as shown in Table 1 were obtained.

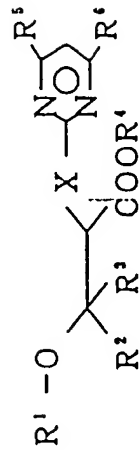
The title compounds (I) obtained as described above are shown in Table 1.

Table 1



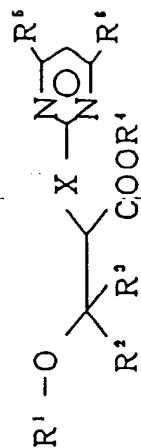
Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	X	Physical property
1	CH ₃	H	CH ₃	CH ₃	OCH ₃	OCH ₃	O	m.p. 83~85 °C
2	"	"	"	C ₂ H ₅	"	"	"	
3	"	"	"	H	"	"	"	m.p. 127~129 °C
4	"	"	"	C ₂ H ₅	"	"	S	
5	"	"	"	H	"	"	"	m.p. 84~86 °C
6	"	"	"	C ₂ H ₅	CH ₃	"	O	
7	"	"	"	H	"	"	"	
8	"	"	"	C ₂ H ₅	"	CH ₃	"	

Table 1 (Contd.)



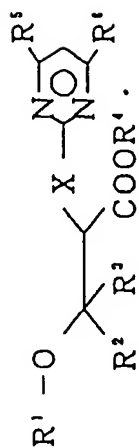
Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	X	Physical property
9	CH ₃	H	CH ₃	H	CH ₃	CH ₃	O	
10	"	"	"	C ₂ H ₅	Cl	OCH ₃	"	
11	"	"	"	H	"	"	"	
12	"	"	"	C ₂ H ₅	CF ₃	"	"	
13	"	"	"	H	"	"	"	
14	"	"	C ₂ H ₅	C ₂ H ₅	OCH ₃	"	"	
15	"	"	"	H	"	"	"	
16	"	"	n-C ₃ H ₇	CH ₃	"	"	"	n _D ^{24.5} 1.4520

Table 1 (Contd.)



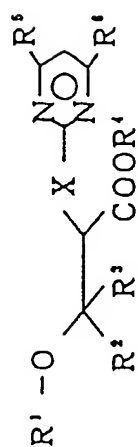
Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	X	Physical property
17	CH ₃	H	n-C ₃ H ₇	H	OCH ₃	OCH ₃	O	m.p. 92~94 °C
18	"	"	i-C ₃ H ₇	CH ₃	"	"	"	n _D ²⁵ 1.4868
19	"	"	"	H	"	"	"	
20	"	CH ₃	CH ₃	C ₂ H ₅	"	"	"	m.p. 89~91 °C
21	"	"	"	H	"	"	"	m.p. 143~145 °C
22	"	"	"	i-C ₃ H ₇	"	"	"	m.p. 73~75 °C
23	"	"	C ₂ H ₅	C ₂ H ₅	"	"	"	m.p. 70~71 °C
24	"	"	"	H	"	"	"	m.p. 138~140 °C

Table 1 (Contd.)



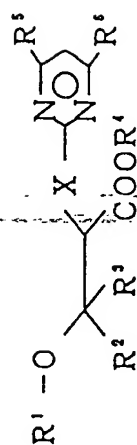
Compound No.	R^1	R^2	R^3	R^4	R^5	R'	X	Physical property
25	CH_3	CH_3		C_2H_5	OCH_3	OCH_3	O	
26	"	"	"	H	"	"	"	
27	"	C_2H_5	C_2H_5	C_2H_5	"	"	"	m.p. 66~67 °C
28	"	"	"	H	"	"	"	m.p. 163~165 °C
29	"			C_2H_5	"	"	"	m.p. 109~110 °C
30	"	"	"	H	"	"	"	m.p. 131~133 °C
31	C_2H_5	H	CH_3	CH_3	"	"	"	Oily product
32	"	"	"	C_2H_5	"	"	"	m.p. 56~58 °C

Table 1 (Contd.)



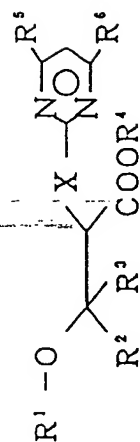
Compound No.	R ¹	R ²	R ³	R ¹	R ⁵	R ⁶	X	Physical property
33	C ₂ H ₅	H	CH ₃	H	OCH ₃	OCH ₃	O	m.p. 111~113 °C
34	"	"	"	C ₂ H ₅	"	"	S	Oily product
35	"	"	"	H	"	"	"	n _D ²⁰ 1.5238
36	"	"	"	C ₂ H ₅	CH ₃	"	O	
37	"	"	"	H	"	"	"	
38	"	"	"	C ₂ H ₅	"	CH ₃	"	
39	"	"	"	H	"	"	"	
40	"	"	"	C ₂ H ₅	Cl	OCH ₃	O	

Table 1 (Contd.)



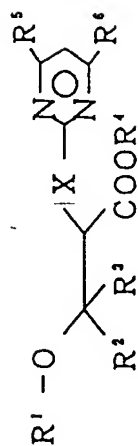
Compound No.	R^1	R^2	R^3	R^4	R^5	R^6	X	Physical property
41	C_2H_5	H	CH_3	H	Cl	OCH_3	O	
42	"	"	"	C_2H_5	CF_3	"	"	
43	"	"	"	H	"	"	"	
44	"	"	C_2H_5	C_2H_5	OCH_3	"	"	n_D^{22} : 1.4704
45	"	"	"	H	"	"	"	m.p. 121~123 °C
46	"	"	n- C_3H_7	C_2H_5	"	"	"	n_D^{24} : 1.4798
47	"	"	"	H	"	"	"	m.p. 123~125 °C
48	"	"	i- C_3H_7	C_2H_5	"	"	"	n_D^{24} : 1.4628

Table 1 (Contd.)



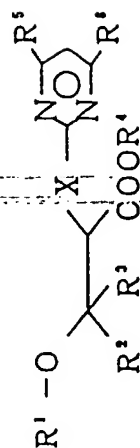
Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	X	Physical property
49	C ₂ H ₅	H	i-C ₃ H ₇	H	OCH ₃	OCH ₃	O	m. p. 115~117 °C
50	"	CH ₃	CH ₃	C ₂ H ₅	"	"	"	m. p. 81~83 °C
51	"	"	"	t-C ₄ H ₉	"	"	"	
52	"	"	"	H	"	"	"	m. p. 121~122 °C
53	"	"	"	C ₂ H ₅	"	"	S	Oily product
54	"	"	"	H	"	"	"	m. p. 96~98 °C
55	"	"	"	C ₂ H ₅	CH ₃	"	O	²³ . ° n _D 1.4812
56	"	"	"	H	"	"	"	m. p. 130~132 °C

Table 1 (Contd.)



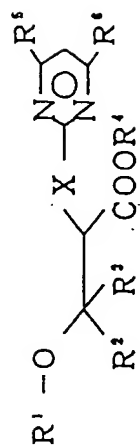
Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	X	Physical property
57	C ₂ H ₅	CH ₃	CH ₃	C ₂ H ₅	CH ₃	CH ₃	O	
58	"	"	"	H	"	"	"	
59	"	"	"	C ₂ H ₅	Cl	OCH ₃	"	n _D ²⁰ 1.4858
60	"	"	"	H	"	"	"	
61	"	"	"	C ₂ H ₅	CF ₃	"	"	n _D ²⁵ 1.4470
62	"	"	"	H	"	"	"	m.p. 73~75 °C
63	"	"	C ₂ H ₅	C ₂ H ₅	OCH ₃	"	"	m.p. 47~48 °C
64	"	"	"	H	"	"	"	m.p. 83~85 °C

Table 1 (Contd.)



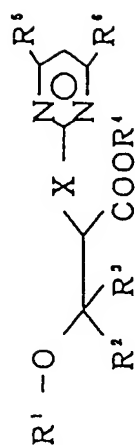
Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	X	Physical property
73	n-C ₃ H ₇	H	CH ₃	C ₂ H ₅	OCH ₃	OCH ₃	O	n _D ²⁵ 1.4776
74	"	"	"	n-C ₃ H ₇	"	"	"	n _D ²⁵ 1.4764
75	"	"	"	H	"	"	"	n _D ²⁵ 1.4866
76	"	"	n-C ₃ H ₇	C ₂ H ₅	"	"	"	n _D ²⁵ 1.4799
77	"	"	"	H	"	"	"	m.p. 129~131 °C
78	"	"	i-C ₃ H ₇	n-C ₃ H ₇	"	"	"	n _D ²⁷ 1.4718
79	"	CH ₃	CH ₃	C ₂ H ₅	"	"	"	m.p. 51~53 °C
80	"	"	"	t-C ₄ H ₉	"	"	"	m.p. 66~68 °C

Table 1 (Contd.)



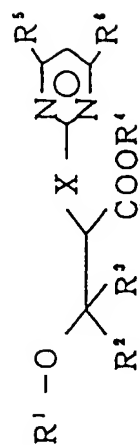
Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	X	Physical property
81	n-C ₃ H ₇	CH ₃	CH ₃	C ₂ H ₅	CF ₃	OCH ₃	O	n _D ²⁴ : 1.4480
82	"	"	"	H	"	"	"	m.p. 78~80 °C
83	"	"	"	"	OCH ₃	"	"	m.p. 99~101 °C
84	i-C ₃ H ₇	H	"	C ₂ H ₅	"	"	"	n _D ²⁴ : 1.4732
85	"	"	"	i-C ₃ H ₇	"	"	"	m.p. 87~89 °C
86	"	"	"	t-C ₄ H ₉	"	"	"	
87	"	"	"	H	"	"	"	m.p. 106~108 °C
88	"	"	"	C ₂ H ₅	"	"	S	n _D ²⁷ : 1.5050

Table 1 (Contd.)



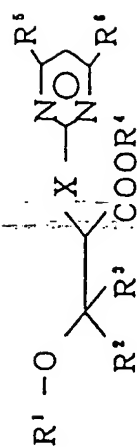
Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	X	Physical property
89	i-C ₃ H ₇	H	CH ₃	C ₂ H ₅	CH ₃	OCH ₃	O	n _D ²⁵ 1.4727
90	"	"	"	H	"	"	"	m.p. 117~120 °C
91	"	"	"	C ₂ H ₅	"	CH ₃	"	
92	"	"	"	H	"	"	"	
93	"	"	"	C ₂ H ₅	Cl	OCH ₃	"	n _D ²⁵ 1.4918
94	"	"	"	H	"	"	"	yellowish oily product
95	"	"	"	C ₂ H ₅	CF ₃	"	"	
96	"	"	"	H	"	"	"	

Table 1 (Contd.)



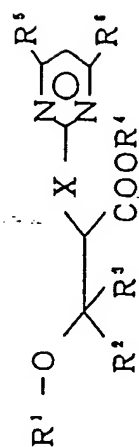
Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	X	Physical property
97	i-C, H ₇	H	C ₂ H ₅	C ₂ H ₅	OCH ₃	OCH ₃	O	n _D ²⁵ 1.4675
98	"	"	n-C ₂ H ₇	"	"	"	"	n _D ²⁵ 1.4742
99	"	"	"	H	"	"	"	m.p. 129~131 °C
100	"	"	i-C ₂ H ₇	C ₂ H ₅	"	"	"	n _D ²⁵ 1.4688
101	"	"	"	i-C ₂ H ₇	"	"	"	n _D ²⁵ 1.4508
102	"	"	"	H	"	"	"	Oily product
103	"	CH ₃	CH ₃	C ₂ H ₅	"	"	"	m.p. 64~66 °C
104	"	"	"	H	"	"	"	Measurement impossible

Table 1 (Contd.)



Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	X	Physical property
105	i-C ₃ H ₇			C ₂ H ₅	OCH ₃	OCH ₃	O	²⁷ n _D ²⁰ 1.4909
106	CH ₂ =CH-CH ₂ -	H	CH ₃	"	"	"	"	²⁵ n _D ²⁰ 1.4879
107	"	"	"	H	"	"	"	m.p. 84~86 °C
108	"	"	"	C ₂ H ₅	"	"	S	
109	"	"	"	H	"	"	"	²³ n _D ²⁰ 1.5277
110	"	CH ₃	"	C ₂ H ₅	"	"	O	m.p. 63~64 °C
111	"	"	"	H	"	"	"	m.p. 65~67 °C
112	"	"	"	C ₂ H ₅	"	"	S	

Table 1 (Contd.)




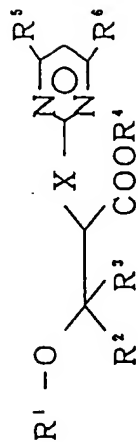
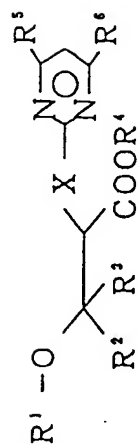
Compound No.	R^1	R^2	R^3	R^4	R^5	R^6	X	Physical property
113	$CH_2=CH-CH_2-$	CH_3	CH_3	H	OCH_3	OCH_3	S	m.p. 83~84 °C
114	"			C_2H_5	"	"	O	m.p. 53~55 °C
115	"	"	"	H	"	"	"	m.p. 94~95 °C
116	$HC\equiv C-CH_2-$	H	CH_3	C_2H_5	"	"	"	$n_D^{27.0}$ 1.4911
117	"	"	"	H	"	"	"	$n_D^{24.4}$ 1.5104
118	"	CH_3	"	C_2H_5	"	"	"	$n_D^{21.4}$ 1.4960
119	"	"	"	$-CH_2C\equiv CH$	"	"	"	$n_D^{21.4}$ 1.5020
120	"	"	"	H	"	"	"	m.p. 136~137 °C

Table 1 (Contd.)



Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	X	Physical property
121	HC≡C-CH ₂ -	CH ₃	CH ₃	C ₂ H ₅	OCH ₃	OCH ₃	S	
122	"	"	"	H	"	"	"	m.p. 81~83 °C
123	n-C ₄ H ₉	H	"	C ₂ H ₅	"	"	O	m.p. 66~68 °C
124	"	"	"	n-C ₄ H ₉	"	"	"	^{24.0} n _D ²⁰ 1.4750
125	"	CH ₃	"	C ₂ H ₅	"	"	"	m.p. 53~55 °C
126	"	"	"	H	"	"	"	m.p. 103~104 °C
127	sec-C ₄ H ₉	H	"	C ₂ H ₅	"	"	"	Measurement impossible
128	"	"	"	sec-C ₄ H ₉	"	"	"	^{25.0} n _D ²⁰ 1.4738

Table 1 (Contd.)




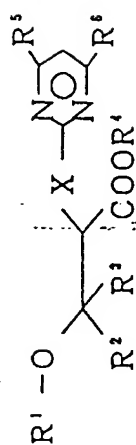
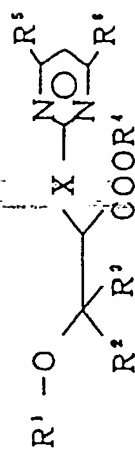
Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	X	Physical property
129	sec-C ₄ H ₉	CH ₃	CH ₃	C ₂ H ₅	OCH ₃	OCH ₃	O	
130	"	"	"	H	"	"	"	
131	i-C ₄ H ₉	H	"	C ₂ H ₅	"	"	"	
132	"	"	"	H	"	"	"	
133	"	CH ₃	"	C ₂ H ₅	"	"	"	
134	"	"	"	H	"	"	"	
135	 -CH ₂ -	H	"	C ₂ H ₅	"	"	"	^{23.6} n _D 1.4921
136	"	"	"	H	"	"	"	

Table 1 (Contd.)



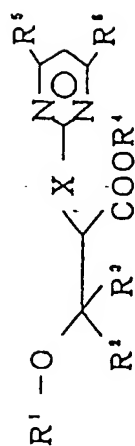
Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	X	Physical property
137		CH ₃	CH ₃	C ₂ H ₅	OCH ₃	OCH ₃	O	m.p. 85~87 °C
138	"	"	"	H	"	"	"	m.p. 105~106 °C
139		H	"	C ₂ H ₅	"	"	"	²⁴ n _D ²⁰ 1.4867
140	"	"	"	H	"	"	"	²⁵ n _D ²⁰ 1.5020
141	"	CH ₃	"	C ₂ H ₅	"	"	"	m.p. 74~77 °C
142	"	"	"	H	"	"	"	m.p. 118~119 °C
143		H	"	C ₂ H ₅	"	"	"	Oily product
144	CF ₃ CH ₂ -	"	"	"	"	"	"	²⁶ n _D ²⁰ 1.4552

Table 1 (Contd.)



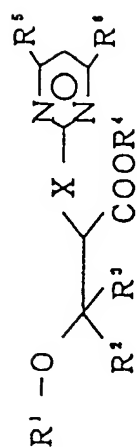
Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	X	Physical property
145	CF ₃ CH ₂ -	H	CH ₃	H	OCH ₃	OCH ₃	O	
146	"	CH ₃	"	C ₂ H ₅	"	"	"	m.p. 54~56 °C
147	"	"	"	H	"	"	"	m.p. 128~130 °C
148	$\left[\begin{array}{c} \text{H}_3\text{C}_2-\text{O} \\ \text{H}_3\text{C}-\text{C}-\text{CH}_2-\text{COO}^- \end{array} \right]$		$\left[\begin{array}{c} \text{O} \\ \text{N} \end{array} \right]$	$\left[\begin{array}{c} \text{OCH}_3 \\ \text{OCH}_3 \end{array} \right]$	⊕ Na			m.p. 200 °C or higher
149	$\left[\begin{array}{c} \text{H}_3\text{C}_2-\text{O} \\ \text{H}_3\text{C}-\text{C}-\text{CH}_2-\text{COO}^- \end{array} \right]$		$\left[\begin{array}{c} \text{O} \\ \text{N} \end{array} \right]$	$\left[\begin{array}{c} \text{OCH}_3 \\ \text{OCH}_3 \end{array} \right]$	⊕ NH ₄			

Table 1 (Contd.)



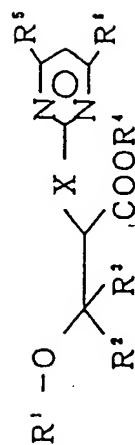
Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	X	Physical property
150	$\left[\begin{array}{c} H_7C_3-O \\ \\ H_3C-CH_2-COO^- \end{array} \right]$	$\left[\begin{array}{c} OCH_3 \\ \\ \text{---} \text{C} \text{---} \text{C} \text{---} \\ \quad \\ O \quad OCH_3 \end{array} \right]$	$\left[\begin{array}{c} OCH_3 \\ \\ \text{---} \text{C} \text{---} \text{C} \text{---} \\ \quad \\ O \quad OCH_3 \end{array} \right]$	$\left[\begin{array}{c} OCH_3 \\ \\ \text{---} \text{C} \text{---} \text{C} \text{---} \\ \quad \\ O \quad OCH_3 \end{array} \right]$	$\left[\begin{array}{c} OCH_3 \\ \\ \text{---} \text{C} \text{---} \text{C} \text{---} \\ \quad \\ O \quad OCH_3 \end{array} \right]$	$\left[\begin{array}{c} OCH_3 \\ \\ \text{---} \text{C} \text{---} \text{C} \text{---} \\ \quad \\ O \quad OCH_3 \end{array} \right]$	$\left[\begin{array}{c} OCH_3 \\ \\ \text{---} \text{C} \text{---} \text{C} \text{---} \\ \quad \\ O \quad OCH_3 \end{array} \right]$	m.p. 200 °C or higher
151	$\left[\begin{array}{c} H_7C_3-O \\ \\ H_3C-CH_2-COO^- \end{array} \right]$	$\left[\begin{array}{c} OCH_3 \\ \\ \text{---} \text{C} \text{---} \text{C} \text{---} \\ \quad \\ O \quad OCH_3 \end{array} \right]$	$\left[\begin{array}{c} OCH_3 \\ \\ \text{---} \text{C} \text{---} \text{C} \text{---} \\ \quad \\ O \quad OCH_3 \end{array} \right]$	$\left[\begin{array}{c} OCH_3 \\ \\ \text{---} \text{C} \text{---} \text{C} \text{---} \\ \quad \\ O \quad OCH_3 \end{array} \right]$	$\left[\begin{array}{c} OCH_3 \\ \\ \text{---} \text{C} \text{---} \text{C} \text{---} \\ \quad \\ O \quad OCH_3 \end{array} \right]$	$\left[\begin{array}{c} OCH_3 \\ \\ \text{---} \text{C} \text{---} \text{C} \text{---} \\ \quad \\ O \quad OCH_3 \end{array} \right]$	$\left[\begin{array}{c} OCH_3 \\ \\ \text{---} \text{C} \text{---} \text{C} \text{---} \\ \quad \\ O \quad OCH_3 \end{array} \right]$	m.p. 122~124 °C
152	CH ₃ -C≡C-CH ₂ -	CH ₃	CH ₃	CH ₃	C ₂ H ₅	OCH ₃	O	m.p. 79~82 °C
153	"	"	"	"	H	"	"	m.p. 118~121 °C
154	C ₂ H ₅	"	"	"	C ₂ H ₅	F	"	n _D ²⁰ 1.4814

Table 1 (Contd.)



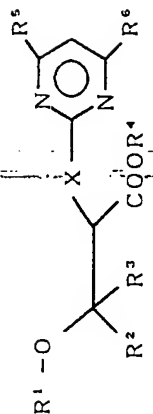
Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	X	Physical property
155	C ₂ H ₅	CH ₃	CH ₃	H	F	OCH ₃	"	n _D ²⁰ 1.4816
156	CH ₃	"	"	"	OCH ₃	"	S	m.p. 97~99 °C
157	C ₂ H ₅	"	"	"	CH ₃	CH ₃	"	m.p. 106~108 °C
158	n-C ₃ H ₇	"	"	"	OCH ₃	OCH ₃	"	m.p. 88~89 °C
159	C ₂ H ₅ CH ₂ -	"	"	C ₂ H ₅	"	"	O	n _D ²² 1.4902
160	"	"	"	H	"	"	"	n _D ²¹ 1.4840
161	"	"	"	"	"	"	S	m.p. 102~103 °C
162	C ₂ H ₅ CH ₂ - CH ₃	H	"	C ₂ H ₅	"	"	O	n _D ²² 1.4895

Table 1 (Contd.)



Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	X	Physical property
163	ClCH ₂ CH- CH ₃	H	CH ₃	H	OCH ₃	OCH ₃	O	m.p. 98~101 °C
164	"	CH ₃	"	C ₂ H ₅	"	"	"	^{22.0} n _D ²⁰ 1.4888
165	"	"	"	H	"	"	"	^{22.0} n _D ²⁰ 1.4920
166	BrCH ₂ CH ₂ -	"	"	C ₂ H ₅	"	"	"	m.p. 60~63 °C
167	"	"	"	H	"	"	"	Oily product
168	CF ₃ >CH- CF ₃	H	"	C ₂ H ₅	"	"	"	^{24.2} n _D ²⁰ 1.4780
169	HC≡C-CH ₂ -	"	"	H	"	"	S	^{24.6} n _D ²⁰ 1.5218
170	CNCH ₂ CH ₂ -	CH ₃	CH ₃	C ₂ H ₅	"	"	O	^{20.0} n _D ²⁰ 1.4912

Table 1 (Contd.)



Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	X	Physical property
171	CNCH ₂ CH ₂ -	CH ₃	CH ₃	H	OCH ₃	OCH ₃	O	m.p. 110~112 °C
172	"	"	"	"	"	"	S	m.p. 116~117 °C
173	i-C ₃ H ₇ -	"	"	"	"	"	"	m.p. 97~98 °C
174	CF ₂ HCH ₂ -	"	"	C ₂ H ₅	"	"	O	n _D ^{24.0} 1.4601
175	"	"	"	H	"	"	"	n _D ^{25.1} 1.4819
176	"	"	"	"	"	"	S	n _D ^{24.5} 1.4968

Example 2

(1) Preparation of granule

8 parts by weight of Compound 21 was uniformly mixed with 30 parts by weight of bentonite, 59 parts

by weight of talc, 1 part by weight of Neopelex powder (trade name, produced by Kao K.K.) and 2 parts by weight of sodium lignosulfonate, and then the mixture was kneaded with addition of a small amount of water, followed by granulation and drying, to obtain a granule.

(2) Preparation of wettable powder

50 parts by weight of Compound 52 was uniformly mixed with 46 parts by weight of kaolin, 2 parts by weight of Neopelex powder (trade name, produced by Kao K.K.) and 2 parts by weight of Demol N (trade name, produced by Kao K.K.), and then the mixture was pulverized to obtain a wettable powder.

(3) Preparation of emulsion

30 parts by weight of Compound 111 was added to 60 parts by weight of xylene, 5 parts by weight of dimethylformamide and 5 parts by weight of Sorpol 3005X (trade name, produced by Toho Kagaku Kogyo) and uniformly mixed to be dissolved therein to obtain an emulsion.

(4) Preparation of dust

5 parts by weight of Compound 120 was uniformly mixed with 50 parts by weight of talc and 45 parts by weight of clay to obtain a dust.

Example 3

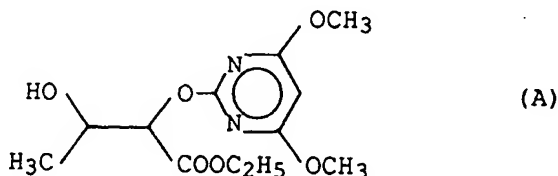
(1). Herbicidal test for paddy field

Wagner pots, each having an area of 1/5000 are, were packed with Ube soil (alluvial soil) and planted with seeds or tubers of barnyardgrass, arrowhead, bulrush and flatstage. Then, the pots were filled with water to a depth of 3 cm.

Each wettable powder of the desired compounds (I) shown in Table 1 prepared in accordance with Example 2 was diluted with water and subjected to dropwise addition treatment by using pipet so that an effective concentration of the compound (I) in each herbicide became 20 g/are at 1 leaf stage of barnyardgrass. These plants were controlled in a glass house at an average temperature of 25 °C for 3 weeks, and then herbicidal effects thereof were investigated.

The herbicidal effects are evaluated according to the 6 ranks (0: None (normal development), 1: Less damaged, 2: Slightly damaged, 3: Moderately damaged, 4: Severely damaged and 5: All killed) as compared with non-treated district.

As a comparative compound, Compound A (Compound No. 76 disclosed in Japanese Provisional Patent Publication No. 85262/1990) shown below was used.



The results are shown in Table 2.

Table 2

Compound	Kind of weed			
	Barnyard-grass	Flatstage	Arrowhead	Bulrush
21	5	5	5	5
52	5	5	5	5
111	5	5	5	5
A	0	0	0	0

(2) Soil treatment test for upland field

Wagner pots, each having an area of 1/5000 are, were packed with Ube soil (alluvial soil), and then each seed of cotton, crabgrass, barnyardgrass, common lambsquarter, livid amaranthus and morning glory were planted and covered with soil.

Each wettable powder of the desired compounds (I) shown in Table 1 prepared in accordance with Example 2 was diluted with water and uniformly sprayed on the surface of each soil so that an effective concentration of the compound (I) in each herbicide became 20 g/are. These plants were controlled in a glass house at an average temperature of 25 °C for 3 weeks, and then herbicidal effects thereof were investigated. The herbicidal effects were evaluated according to the evaluation method described in Test example (1), and the results are shown in Table 3 (as a comparative compound, Compound A shown above was used).

Table 3

Compound	Kind of weed					Cotton
	Crab-grass	Barn-yard-grass	Common lambs-quarter	Livid amaranthus	Morning glory	
1	4	4	5	5	2	0
3	3	4	5	5	3	0
21	5	5	5	5	4	1
33	4	5	5	5	3	0
52	5	5	5	5	5	2
106	4	3	5	5	4	1
107	4	3	5	5	3	1
111	5	5	5	5	4	1
119	4	4	5	5	3	0
120	4	4	5	5	4	0
A	2	2	1	1	0	0

(3) Foliar spread test for upland field

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Wagner pots, each having an area of 1/5000 are, were packed with volcanic ash soil and then each seed of crabgrass, barnyardgrass, common lambsquarter, livid amaranthus and morning glory was planted, covered with soil and grown for 2 weeks.

Each wettable powder of the desired compounds (I) shown in Table 1 prepared in accordance with Example 2 was diluted to 2000 ppm with water containing a surfactant (0.05 %) and then uniformly sprayed on the above respective plants. After these plants were controlled in a glass house at an average temperature of 25 °C for 3 weeks, the herbicidal effects thereof were investigated.

The herbicidal effects were evaluated according to the evaluation method described in Test example (1), and the results are shown in Table 4 (as a comparative compound, Compound A shown above was used).

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Table 4

Com- pound	Kind of weed				
	Crab- grass	Barnyard- grass	Common lambs- quarter	Livid amaran- thus	Morning glory
3	4	5	5	5	4
5	5	5	5	5	4
15	3	4	5	5	3
21	5	5	5	5	4
33	4	5	5	5	3
34	5	5	5	5	4
35	5	5	5	5	5
52	5	5	5	5	5
54	5	5	5	5	5
106	5	5	5	5	4
107	5	5	5	5	5
109	5	5	5	5	3
111	5	5	5	5	5
113	5	5	5	5	5
119	4	5	5	5	4
120	5	5	5	5	5
122	5	5	5	5	5
153	4	4	4	5	3
155	5	5	5	5	5
156	5	5	5	5	5
157	5	4	5	5	3
158	5	5	5	5	5
160	5	5	5	5	5
161	5	5	5	5	5
163	5	5	5	5	3
165	5	5	5	5	5
166	3	5	5	5	3
167	5	5	5	5	5

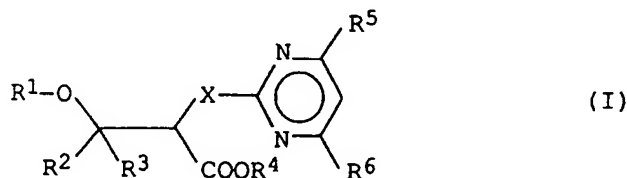
Table 4 (contd.)

Com- pound	Kind of weed				
	Crab- grass	Barnyard- grass	Common lambs- quarter	Livid amaran- thus	Morning glory
168	5	5	5	5	5
169	5	5	5	5	5
170	5	5	5	5	5
171	5	5	5	5	5
172	5	5	5	5	5
A	2	3	1	1	1

The novel 3-alkoxyalkanoic acid derivative of the present invention has high selectivity to annual and perennial weeds, and also shows excellent herbicidal effect (particularly effective on annual grass weeds and broad-leaved weeds).

Claims

1. A 3-alkoxyalkanoic acid compound represented by the following formula (I):

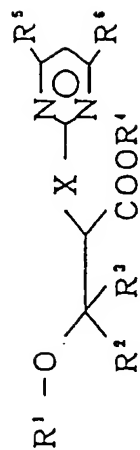


wherein R¹ represents a lower alkyl group, a lower alkenyl group, a lower alkynyl group, a cycloalkyl group, a halo-lower alkyl group or a cyano-lower alkyl group; R² represents hydrogen atom or a lower alkyl group; R³ represents a lower alkyl group or a cycloalkyl group; or R² and R³ are mutually bonded to represent a cycloalkyl group; R⁴ represents hydrogen atom, a lower alkyl group or a lower alkynyl group; R⁵ represents a lower alkoxy group, a lower alkyl group, a halogen atom or a halo-lower alkyl group; R⁶ represents a lower alkoxy group or a lower alkyl group; and X represents oxygen atom or sulfur atom, or an alkali addition salt thereof.

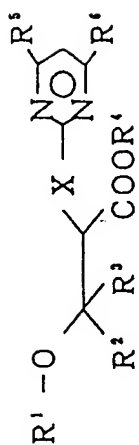
2. The 3-alkoxyalkanoic acid compound according to Claim 1, wherein R¹ is a straight or branched alkyl group having 1 to 6 carbon atoms which may be substituted by a cycloalkyl group having 3 to 8 carbon atoms, a straight or branched alkenyl group having 2 to 6 carbon atoms, a straight or branched alkynyl group having 2 to 6 carbon atoms, a cycloalkyl group having 3 to 8 carbon atoms and a straight or branched cyanoalkyl group having 2 to 6 carbon atoms; R² is hydrogen atom or a straight or branched alkyl group having 1 to 5 carbon atoms; R³ is a straight or branched alkyl group having 1 to 6 carbon atoms; or R² and R³ may be mutually bonded to form a cycloalkyl group having 3 to 8 carbon atoms; R⁴ is hydrogen atom, a straight or branched alkyl group having 1 to 5 carbon atoms or a straight or branched alkynyl group having 2 to 5 carbon atoms; R⁵ is a straight or branched alkoxy group having 1 to 5 carbon atoms, a straight or branched alkyl group having 1 to 5 carbon atoms, fluorine atom, chlorine atom, bromine atom, iodine atom or a straight or branched haloalkyl group having 1 to 5 carbon atoms; R⁶ is a straight or branched alkoxy group having 1 to 5 carbon atoms or a straight or branched alkyl group having 1 to 5 carbon atoms; R⁷ is a straight or branched alkyl group having 1 to 5

carbon atoms, a substituted phenyl group or a straight or branched halo-lower alkyl group having 1 to 5 carbon atoms; and X is oxygen atom or sulfur atom.

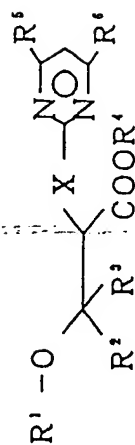
3. The 3-alkoxyalkanoic acid compound according to Claim 2, wherein R¹ is an alkyl group having 1 to 4 carbon atoms, a straight or branched alkyl group having 1 to 3 carbon atoms which may be substituted by a cycloalkyl group having 3 to 5 carbon atoms, a straight or branched alkenyl group having 2 to 5 carbon atoms, a straight or branched alkynyl group having 2 to 5 carbon atoms, a cycloalkyl group having 3 to 6 carbon atoms, a haloalkyl group having 2 to 4 carbon atoms or a straight or branched cyanoalkyl group having 2 to 4 carbon atoms.
4. The 3-alkoxyalkanoic acid compound according to Claim 2, wherein R¹ is methyl group, ethyl group, n-propyl group, i-propyl group, n-butyl group, i-butyl group, sec-butyl group, t-butyl group, cyclopropylmethyl group, cyclobutylmethyl group, propenyl group, propynyl group or cyclohexyl group.
5. The 3-alkoxyalkanoic acid compound according to Claim 1, wherein said compound is at least one selected from the group consisting of:



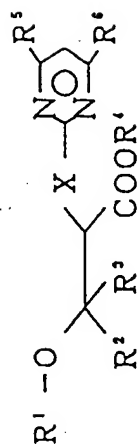
Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	X	Physical property
1	CH ₃	H	CH ₃	CH ₃	OCH ₃	OCH ₃	O	m.p. 83~85 °C
2	"	"	"	C ₂ H ₅	"	"	"	
3	"	"	"	H	"	"	"	m.p. 127~129 °C
4	"	"	"	C ₂ H ₅	"	"	S	
5	"	"	"	H	"	"	"	m.p. 84~86 °C
6	"	"	"	C ₂ H ₅	CH ₃	"	O	
7	"	"	"	H	"	"	"	
8	"	"	"	C ₂ H ₅	"	CH ₃	"	



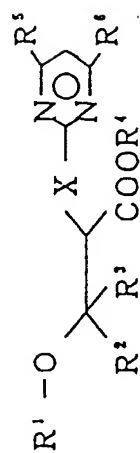
Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	X	Physical property
9	CH ₃	H	CH ₃	H	CH ₃	CH ₃	O	
10	"	"	"	C ₂ H ₅	Cl	OCH ₃	"	
11	"	"	"	H	"	"	"	
12	"	"	"	C ₂ H ₅	CF ₃	"	"	
13	"	"	"	H	"	"	"	
14	"	"	C ₂ H ₅	C ₂ H ₅	OCH ₃	"	"	
15	"	"	"	H	"	"	"	
16	"	"	n-C ₄ H ₉	CH ₃	"	"	"	n _D ²⁰ 1.4520



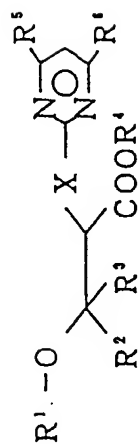
Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	X	Physical property
17	CH ₃	H	n-C ₃ H ₇	H	OCH ₃	OCH ₃	O	m.p. 92~94 °C
18	"	"	i-C ₃ H ₇	CH ₃	"	"	"	n _D ²⁰ 1.4868
19	"	"	"	H	"	"	"	
20	"	CH ₃	CH ₃	C ₂ H ₅	"	"	"	m.p. 89~91 °C
21	"	"	"	H	"	"	"	m.p. 143~145 °C
22	"	"	"	i-C ₃ H ₇	"	"	"	m.p. 73~75 °C
23	"	"	C ₂ H ₅	C ₂ H ₅	"	"	"	m.p. 70~71 °C
24	"	"	"	H	"	"	"	m.p. 138~140 °C



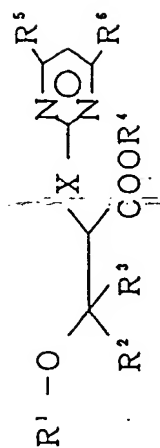
Compound No.	R^1	R^2	R^3	R^4	R^5	R^6	X	Physical property
25	CH_3	CH_3		C_2H_5	OCH_3	OCH_3	O	
26	"	"	"	H	"	"	"	
27	"	C_2H_5	C_2H_5	C_2H_5	"	"	"	m.p. 66~67 °C
28	"	"	"	H	"	"	"	m.p. 163~165 °C
29	"		"	C_2H_5	"	"	"	m.p. 109~110 °C
30	"	"	"	H	"	"	"	m.p. 131~133 °C
31	C_2H_5	H	CH_3	CH_3	"	"	"	Oily product
32	"	"	"	C_2H_5	"	"	"	m.p. 56~58 °C



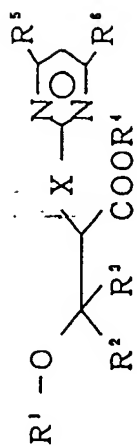
Compound No.	R^1	R^2	R^3	R^4	R^5	R^6	X	Physical property
33	C_2H_5	H	CH_3	H	OCH_3	OCH_3	O	m.p. 111~113 °C
34	"	"	"	C_2H_5	"	"	S	Oily product
35	"	"	"	H	"	"	"	n_D^{20} 1.5238
36	"	"	"	C_2H_5	CH_3	"	O	
37	"	"	"	H	"	"	"	
38	"	"	"	C_2H_5	"	CH_3	"	
39	"	"	"	H	"	"	"	
40	"	"	"	C_2H_5	Cl	OCH_3	O	



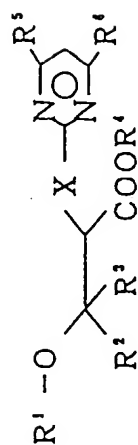
Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	X	Physical property
41	C ₂ H ₅	H	CH ₃	H	Cl	OCH ₃	O	
42	"	"	"	C ₂ H ₅	CF ₃	"	"	
43	"	"	"	H	"	"	"	
44	"	"	C ₂ H ₅	C ₂ H ₅	OCH ₃	"	"	n _D ²⁰ 1.4704
45	"	"	"	H	"	"	"	m.p. 121~123 °C
46	"	"	n-C ₃ H ₇	C ₂ H ₅	"	"	"	n _D ²⁰ 1.4798
47	"	"	"	H	"	"	"	m.p. 123~125 °C
48	"	"	i-C ₃ H ₇	C ₂ H ₅	"	"	"	n _D ²⁰ 1.4628






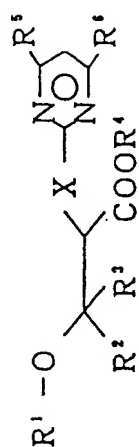
Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	X	Physical property
49	C ₂ H ₅	H	i-C ₃ H ₇	H	OCH ₃	OCH ₃	O	m.p. 115~117 °C
50	"	CH ₃	CH ₃	C ₂ H ₅	"	"	"	m.p. 81~83 °C
51	"	"	"	t-C ₄ H ₉	"	"	"	
52	"	"	"	H	"	"	"	m.p. 121~122 °C
53	"	"	"	C ₂ H ₅	"	"	S	Oily product
54	"	"	"	H	"	"	"	m.p. 96~98 °C
55	"	"	"	C ₂ H ₅	CH ₃	"	O	n _D ²⁰ 1.4812
56	"	"	"	H	"	"	"	m.p. 130~132 °C



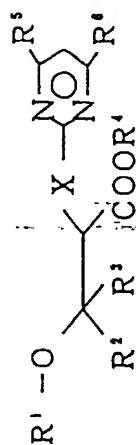
Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	X	Physical property
57	C ₂ H ₅	CH ₃	CH ₃	C ₂ H ₅	CH ₃	CH ₃	O	
58	"	"	"	H	"	"	"	
59	"	"	"	C ₂ H ₅	Cl	OCH ₃	"	n _D ²⁵ 1.4858
60	"	"	"	H	"	"	"	
61	"	"	"	C ₂ H ₅	CF ₃	"	"	n _D ²⁵ 1.4470
62	"	"	"	H	"	"	"	m.p. 73~75 °C
63	"	"	C ₂ H ₅	C ₂ H ₅	OCH ₃	"	"	m.p. 47~48 °C
64	"	"	"	H	"	"	"	m.p. 83~85 °C



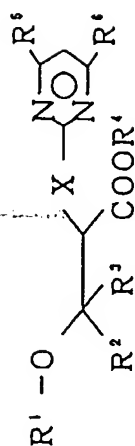
Compound No.	R^1	R^2	R^3	R^4	R^5	R^6	X	Physical property
65	C_2H_5	CH_3		C_2H_5	OCH_3	OCH_3	O	
66	"	"	"	H	"	"	"	
67	"	C_2H_5	C_2H_5	C_2H_5	"	"	"	n_D^{20} 1.4815
68	"	"	"	H	"	"	"	m.p. 131~133 °C
69	"			C_2H_5	"	"	"	m.p. 82~83 °C
70	"	"	"	H	"	"	"	m.p. 156~158 °C
71	"			C_2H_5	"	"	"	m.p. 89~91 °C
72	"	"	"	H	"	"	"	m.p. 173~175 °C



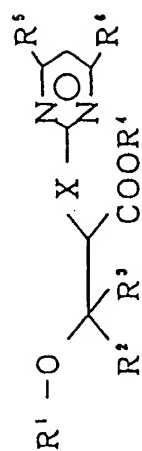
Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	X	Physical property
73	n-C ₃ H ₇	H	CH ₃	C ₂ H ₅	OCH ₃	OCH ₃	O	²⁶ n _D ²⁰ 1.4776
74	"	"	"	n-C ₃ H ₇	"	"	"	²⁶ n _D ²⁰ 1.4764
75	"	"	"	H	"	"	"	²⁵ n _D ²⁰ 1.4866
76	"	"	n-C ₃ H ₇	C ₂ H ₅	"	"	"	²⁴ n _D ²⁰ 1.4799
77	"	"	"	H	"	"	"	m.p. 129~131 °C
78	"	"	i-C ₃ H ₇	n-C ₃ H ₇	"	"	"	²⁷ n _D ²⁰ 1.4718
79	"	CH ₃	CH ₃	C ₂ H ₅	"	"	"	m.p. 51~53 °C
80	"	"	"	t-C ₄ H ₉	"	"	"	m.p. 66~68 °C



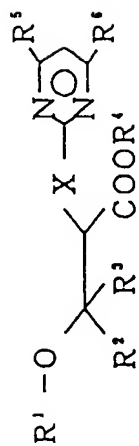
Compound No.	R^1	R^2	R^3	R^4	R^5	R^6	X	Physical property
81	n-C ₃ H ₇	CH ₃	CH ₃	C ₂ H ₅	CF ₃	OCH ₃	O	n_D^{24} 1.4480
82	"	"	"	H	"	"	"	m.p. 78~80 °C
83	"	"	"	"	OCH ₃	"	"	m.p. 99~101 °C
84	i-C ₃ H ₇	H	"	C ₂ H ₅	"	"	"	n_D^{24} 1.4732
85	"	"	"	i-C ₃ H ₇	"	"	"	m.p. 87~89 °C
86	"	"	"	t-C ₄ H ₉	"	"	"	
87	"	"	"	H	"	"	"	m.p. 106~108 °C
88	"	"	"	C ₂ H ₅	"	"	S	n_D^{27} 1.5050




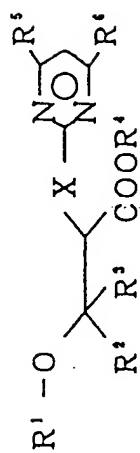
Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	X	Physical property
89	i-C ₃ H ₇	H	CH ₃	C ₂ H ₅	CH ₃	OCH ₃	O	^{23.} n _D ²⁰ 1.4727
90	"	"	"	H	"	"	"	m.p. 117~120 °C
91	"	"	"	C ₂ H ₅	"	CH ₃	"	
92	"	"	"	H	"	"	"	
93	"	"	"	C ₂ H ₅	Cl	OCH ₃	"	^{25.} n _D ²⁰ 1.4918
94	"	"	"	H	"	"	"	Yellowish oily product
95	"	"	"	C ₂ H ₅	CF ₃	"	"	
96	"	"	"	H	"	"	"	




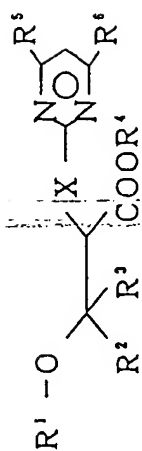
Compound No.	R^1	R^2	R^3	R^4	R^5	R^6	X	Physical property
97	i-C ₃ H ₇	H	C ₂ H ₅	C ₂ H ₅	OCH ₃	OCH ₃	O	n_D^{25} 1.4675
98	"	"	n-C ₃ H ₇	"	"	"	"	n_D^{25} 1.4742
99	"	"	"	H	"	"	"	m.p. 129~131 °C
100	"	"	i-C ₃ H ₇	C ₂ H ₅	"	"	"	n_D^{25} 1.4688
101	"	"	"	i-C ₃ H ₇	"	"	"	n_D^{25} 1.4508
102	"	"	"	H	"	"	"	Oily product
103	"	CH ₃	CH ₃	C ₂ H ₅	"	"	"	m.p. 64~66 °C
104	"	"	"	H	"	"	"	Measurement impossible



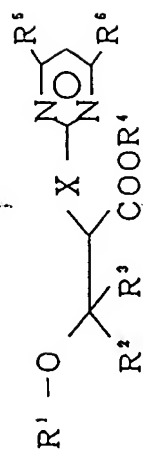
Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	X	Physical property
105	i-C ₃ H ₇			C ₂ H ₅	OCH ₃	OCH ₃	O	n _D ^{21.5} 1.4909
106	CH ₂ =CH-CH ₂ -	H	CH ₃	"	"	"	"	n _D ^{25.0} 1.4879
107	"	"	"	H	"	"	"	m.p. 84~86 °C
108	"	"	"	C ₂ H ₅	"	"	S	
109	"	"	"	H	"	"	"	n _D ^{22.5} 1.5277
110	"	CH ₃	"	C ₂ H ₅	"	"	O	m.p. 63~64 °C
111	"	"	"	H	"	"	"	m.p. 65~67 °C
112	"	"	"	C ₂ H ₅	"	"	S	



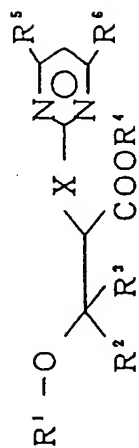
Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	X	Physical property
113	CH ₂ =CH-CH ₂ -	CH ₃	CH ₃	H	OCH ₃	OCH ₃	S	m.p. 83~84 °C
114	"			C ₂ H ₅	"	"	O	m.p. 53~55 °C
115	"	"	"	H	"	"	"	m.p. 94~95 °C
116	HC≡C-CH ₂ -	H	CH ₃	C ₂ H ₅	"	"	"	n _D ^{27.°} 1.4911
117	"	"	"	H	"	"	"	n _D ^{24.°} 1.5104
118	"	CH ₃	"	C ₂ H ₅	"	"	"	n _D ^{21.°} 1.4960
119	"	"	"	-CH ₂ C≡CH	"	"	"	n _D ^{21.°} 1.5020
120	"	"	"	H	"	"	"	m.p. 136~137 °C






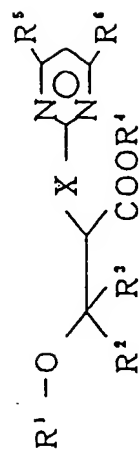
Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	X	Physical property
121	HC≡C-CH ₂ -	CH ₃	CH ₃	C ₂ H ₅	OCH ₃	OCH ₃	S	
122	"	"	"	H	"	"	"	m. p. 81~83 °C
123	n-C ₄ H ₉	H	"	C ₂ H ₅	"	"	O	m. p. 66~68 °C
124	"	"	"	n-C ₄ H ₉	"	"	"	²⁴ . ^o n _D 1.4750
125	"	CH ₃	"	C ₂ H ₅	"	"	"	m. p. 53~55 °C
126	"	"	"	H	"	"	"	m. p. 103~104 °C
127	sec-C ₄ H ₉	H	"	C ₂ H ₅	"	"	"	Measurement impossible
128	"	"	"	sec-C ₄ H ₉	"	"	"	²⁵ . ^o n _D 1.4738



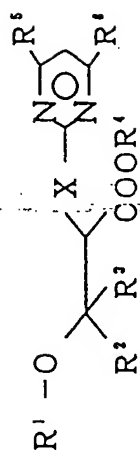
Compound No.	R^1	R^2	R^3	R^4	R^5	R^6	X	Physical property
129	sec-C ₄ H ₉	CH ₃	CH ₃	C ₂ H ₅	OCH ₃	OCH ₃	O	
130	"	"	"	H	"	"	"	
131	i-C ₄ H ₉	H	"	C ₂ H ₅	"	"	"	
132	"	"	"	H	"	"	"	
133	"	CH ₃	"	C ₂ H ₅	"	"	"	
134	"	"	"	H	"	"	"	
135	$\triangle-CH_2-$	H	"	C ₂ H ₅	"	"	"	$n_D^{22} 1.4921$
136	"	"	"	H	"	"	"	



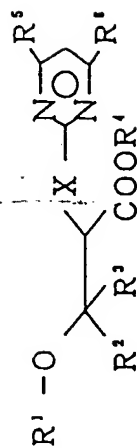
Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	X	Physical property
137	 -CH ₂ -	CH ₃	CH ₃	C ₂ H ₅	OCH ₃	OCH ₃	O	m.p. 85~87 °C
138	"	"	"	H	"	"	"	m.p. 105~106 °C
139	 -CH ₂ -	H	"	C ₂ H ₅	"	"	"	n _D ²⁴ 1.4867
140	"	"	"	H	"	"	"	n _D ²⁵ 1.5020
141	"	CH ₃	"	C ₂ H ₅	"	"	"	m.p. 74~77 °C
142	"	"	"	H	"	"	"	m.p. 118~119 °C
143		H	"	C ₂ H ₅	"	"	"	Oily product
144	CF ₃ CH ₂ -	"	"	"	"	"	"	n _D ²⁶ 1.4552



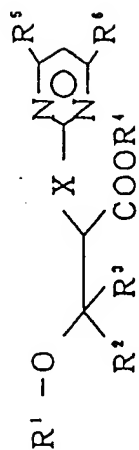
Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	X	Physical property
145	CF ₃ CH ₂ -	H	CH ₃	H	OCH ₃	OCH ₃	O	
146	"	CH ₃	"	C ₂ H ₅	"	"	"	m.p. 54~56 °C
147	"	"	"	H	"	"	"	m.p. 128~130 °C
148	$\left[\begin{array}{c} \text{H}_3\text{C}_2-\text{O} \\ \text{H}_3\text{C} \end{array} \right]$	$\begin{array}{c} \text{CH}_3 \\ \text{CH}_3 \end{array}$	$\begin{array}{c} \text{O} \\ \text{COO}^- \end{array}$	$\left[\begin{array}{c} \text{OCH}_3 \\ \text{OCH}_3 \end{array} \right]$	Na^+			m.p. 200 °C or higher
149	$\left[\begin{array}{c} \text{H}_3\text{C}_2-\text{O} \\ \text{H}_3\text{C} \end{array} \right]$	$\begin{array}{c} \text{CH}_3 \\ \text{CH}_3 \end{array}$	$\begin{array}{c} \text{O} \\ \text{COO}^- \end{array}$	$\left[\begin{array}{c} \text{OCH}_3 \\ \text{OCH}_3 \end{array} \right]$	NH_4^+			



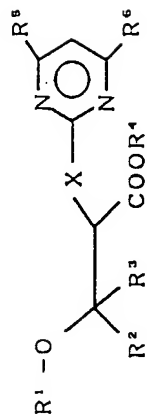
Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	X	Physical property
150	$\left[\begin{array}{c} \text{H}_7 \text{ C}_3 - \text{O} \\ \text{H}_3 \text{ C} \end{array} \right]$	$\left[\begin{array}{c} \text{CH}_3 \\ \text{CH}_3 \end{array} \right]$	$\left[\begin{array}{c} \text{O} \\ \text{COO}^- \end{array} \right]$	$\left[\begin{array}{c} \text{OCH}_3 \\ \text{OCH}_3 \end{array} \right]$	Na^+			m.p. 200 °C or higher
151	$\left[\begin{array}{c} \text{H}_7 \text{ C}_3 - \text{O} \\ \text{H}_3 \text{ C} \end{array} \right]$	$\left[\begin{array}{c} \text{CH}_3 \\ \text{CH}_3 \end{array} \right]$	$\left[\begin{array}{c} \text{O} \\ \text{COO}^- \end{array} \right]$	$\left[\begin{array}{c} \text{OCH}_3 \\ \text{OCH}_3 \end{array} \right]$	NH_3^+	$\text{CH}(\text{CH}_3)_2$		m.p. 122~124 °C
152	$\text{CH}_3 - \text{C} \equiv \text{C} - \text{CH}_2 -$	CH_3	CH_3	$\text{C}_2 \text{ H}_5$	OCH_3	OCH_3	O	m.p. 79~82 °C
153	"	"	"	H	"	"	"	m.p. 118~121 °C
154	$\text{C}_2 \text{ H}_5$	"	"	$\text{C}_2 \text{ H}_5$	F	"	"	n_D^{20} 1.4814



Compound No.	R^1	R^2	R^3	R^4	R^5	R^6	X	Physical property
155	C_2H_5	CH_3	CH_3	H	F	OCH_3	"	n_D^{20} 1.4816
156	CH_3	"	"	"	OCH_3	"	S	m.p. 97~99 °C
157	C_2H_5	"	"	"	CH_3	CH_3	"	m.p. 106~108 °C
158	$n\text{-C}_3\text{H}_7$	"	"	"	OCH_3	OCH_3	"	m.p. 88~89 °C
159	$\text{C}_2\text{H}_5\text{CH}_2\text{CH}_2\text{-}$	"	"	C_2H_5	"	"	O	n_D^{22} 1.4902
160	"	"	"	H	"	"	"	n_D^{21} 1.4840
161	"	"	"	"	"	"	S	m.p. 102~103 °C
162	$\text{C}_2\text{H}_5\text{CH}_2\text{CH-CH}_3$	H	"	C_2H_5	"	"	O	n_D^{23} 1.4895



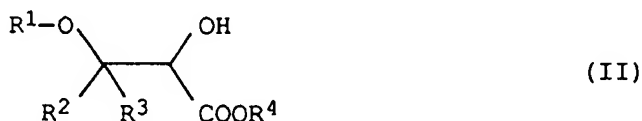
Compound No.	R ¹	R ²	R ³	R ⁴	R ⁵	R ⁶	X	Physical property
163	ClCH ₂ CH- CH ₃	H	CH ₃	H	OCH ₃	OCH ₃	O	m.p. 98~101 °C
164	"	CH ₃	"	C ₂ H ₅	"	"	"	^{23.} n _D ²⁰ 1.4888
165	"	"	"	H	"	"	"	^{23.} n _D ²⁰ 1.4920
166	BrCH ₂ CH ₂ -	"	"	C ₂ H ₅	"	"	"	m.p. 60~63 °C
167	"	"	"	H	"	"	"	Oily product
168	CF ₃ >CH- CF ₃	H	"	C ₂ H ₅	"	"	"	^{24.} n _D ²⁰ 1.4780
169	HC≡C-CH ₂ -	"	"	H	"	"	S	^{24.} n _D ²⁰ 1.5218
170	CNCH ₂ CH ₂ -	CH ₃	CH ₃	C ₂ H ₅	"	"	O	^{20.} n _D ²⁰ 1.4912



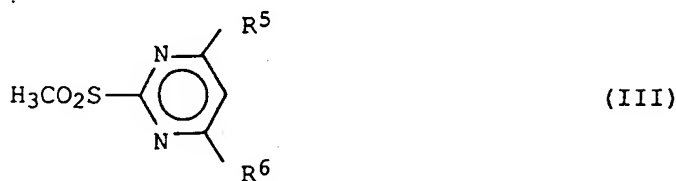
Compound No.	R^1	R^2	R^3	R^4	R^5	R^6	X	Physical property
171	$CNCH_2CH_2-$	CH_3	CH_3	H	OCH_3	OCH_3	O	m.p. 110~112 °C
172	"	"	"	"	"	"	S	m.p. 116~117 °C
173	$i-C_3H_7-$	"	"	"	"	"	"	m.p. 97~98 °C
174	CF_2HCH_2-	"	"	C_2H_5	"	"	O	$n_D^{24.0}$ 1.4601
175	"	"	"	H	"	"	"	$n_D^{25.1}$ 1.4819
176	"	"	"	"	"	"	S	$n_D^{24.5}$ 1.4968

6. The 3-alkoxyalkanoic acid compound according to Claim 5, wherein said compound is at least one selected from Compounds Nos. 1, 2, 6, 8, 10, 12, 14, 16, 18, 20, 22, 23, 25, 27, 29, 31, 32, 36, 38, 40, 42, 44, 46, 48, 50, 51, 55, 57, 59, 61, 63, 65, 67, 69, 71, 73, 74, 76, 78, 79, 80, 81, 84, 85, 86, 89, 91, 93, 95, 97, 98, 100, 101, 103, 105, 106, 110, 114, 116, 118, 119, 123, 124, 125, 127, 128, 129, 131, 133, 135, 137, 139, 141, 143, 144, 146, 152, 154, 159, 162, 164, 166, 168, 170 and 174.

7. A process for preparing the 3-alkoxyalkanoic acid compound represented by the formula (I) according to Claim 1, which comprises reacting a compound represented by the following formula (II):

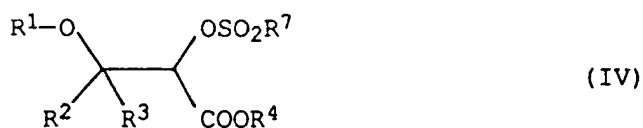


10 wherein R¹, R², R³ and R⁴ each have the same meanings as defined in Claim 1 with a compound represented by the following formula (III):

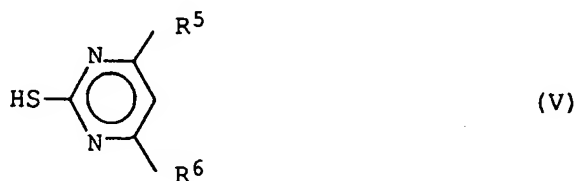


wherein R⁵ and R⁶ each have the same meanings as defined in Claim 1.

- 25 8. The process according to Claim 7, wherein said reaction is carried out in a solvent in the presence of a base at a temperature of the boiling point of the solvent or lower for 1 to 12 hours.
9. The process according to Claim 7, wherein the compound of the formula (III) is used in an amount of 0.5 to 2 mole per mole of the compound of the formula (II).
- 30 10. A process for preparing the 3-alkoxyalkanoic acid compound represented by the formula (I) according to Claim 1, which comprises reacting a compound represented by the following formula (IV):



40 wherein R¹, R², R³ and R⁴ each have the same meanings as defined in Claim 1; and R⁷ represents a lower alkyl group, a substituted phenyl group or a halo-lower alkyl group with a compound represented by the following formula (V):

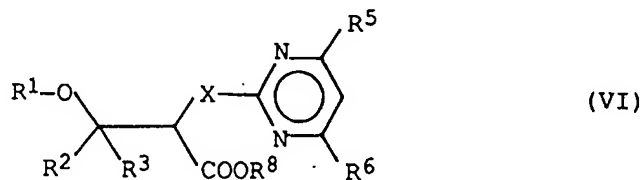


55 wherein R⁵ and R⁶ each have the same meanings as defined in Claim 1.

11. The process according to Claim 10, wherein said reaction is carried out in a solvent in the presence of a base at a temperature in the range of 10 to 80 °C for 1 to 12 hours.

12. The process according to Claim 10, wherein the compound of the formula (V) is used in an amount of 0.5 to 2 mole per mole of the compound of the formula (IV).

13. A process for preparing the 3-alkoxyalkanoic acid compound represented by the formula (I) according to Claim 1, which comprises hydrolyzing a compound represented by the following formula (VI):



15 wherein R¹, R², R³, R⁵, R⁶ and X each have the same meanings as defined in Claim 1; and R⁸ represents a lower alkyl group or a lower alkynyl group.

14. The process according to Claim 13, wherein hydrolysis is carried out in a solvent in the presence of a base at a temperature in the range of 10 to 80 °C for 1 to 12 hours.

15. A herbicide comprising the 3-alkoxyalkanoic acid compound represented by the formula (I) according to Claim 1 or an alkali addition salt thereof as an active ingredient and a herbicidally effective carrier.



European
Patent Office

EUROPEAN SEARCH REPORT

Application Number

EP 91 11 7829

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int. C1.5)
A	EP-A-0 347 811 (KUMIAI CHEMICAL INDUSTRY) * CLAIMS ** -----	1-5,15	C 07 D 239/60 C 07 D 239/52 C 07 D 239/34
P,X	EP-A-0 409 368 (SCHERING) * page 1 - page 26 ** -----	1-15	A 01 N 43/54
			TECHNICAL FIELDS SEARCHED (Int. C1.5)
			C 07 D
The present search report has been drawn up for all claims			
Place of search		Date of completion of search	Examiner
The Hague		27 January 92	FRANCOIS J.C.L.
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